

**$\beta,\gamma$ -EPOXY SULFONES IN ORGANIC SYNTHESIS. PART 2:<sup>†</sup>**  
**PREPARATION OF  $\beta,\gamma$ -BIFUNCTIONALIZED SULFONES**

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**Abstract:** Heteronucleophiles reagents react with  $\beta,\gamma$ -epoxy sulfones 1 at the  $\gamma$ -position in the presence of  $Ti(OPr)_4$ , or magnesium halides to afford regioselectively  $\gamma$ -functionalized  $\beta$ -hydroxy sulfones 2. With soft nucleophiles such as sodium azide, triphenylphosphine, or sodium benzethiolate the reaction takes place in the absence of Lewis acids. In the case of basic reagents such as sodium methoxide or diethylamine the corresponding regioisomers, the  $\beta$ -functionalized  $\gamma$ -hydroxy sulfones 3 were exclusively obtained. The preparation of compounds 3 ( $R^2=H$ ) was also carried out starting from 3-tosyl-2-propen-1-ol (4) by Michael addition of different heteronucleophiles.

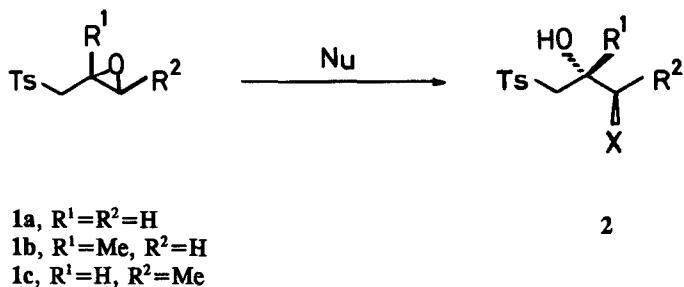
The reactivity of  $\beta,\gamma$ -epoxy sulfones with nucleophilic reagents is based in two types of ring-opening: (a)  $\alpha$ -proton abstraction and  $\beta$ -elimination to afford the corresponding allylic alcohol and (b) nucleophilic attack at the  $\gamma$ -position. The first reaction has been widely used in synthesis of cyclic<sup>1</sup> and acyclic<sup>2</sup>  $\alpha,\beta$ -unsaturated carbonyl compounds, and  $\gamma$ -arylsulfonyl allylic alcohols.<sup>3</sup> The second process is the classic nucleophilic opening of epoxides, which has been only carried out with organomagnesium compounds and copper(I) salts<sup>2,4</sup> to afford  $\beta$ -hydroxy sulfones. Following our studies on the reactivity of  $\beta,\gamma$ -epoxy sulfones<sup>2,3f</sup> we report here the ring-opening of these compounds by reaction with heteroatomic nucleophiles.

## RESULTS AND DISCUSSION

The reaction of  $\beta,\gamma$ -epoxy sulfones 1 with different heteroatomic nucleophiles gave in general  $\gamma$ -functionalized  $\beta$ -hydroxy sulfones 2 resulting from the attack at the  $\gamma$ -position (Scheme 1 and Table 1). Thus, the ring-opening of epoxy sulfones 1 occurred in the presence of  $Ti(OPr)_4$ ,<sup>5</sup> when isopropanol, ammonium thiocyanate, benzenethiol, ethylmagnesium bromide, iodine,<sup>6</sup> and trimethylsilyl cyanide were used

<sup>†</sup>For part 1 see reference 2.

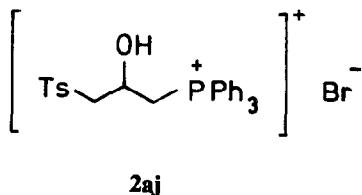
as nucleophiles (Table 1, entries 2, 8-10, 11, 17, 21, and 23, respectively). No reaction was observed with other nucleophiles, such as carboxylic acids or ammonium carboxylates, ammonium chloride, and potassium cyanide in the presence of  $\text{Ti}(\text{OPr})_4$ .<sup>5</sup> Magnesium halogenides<sup>7</sup> are the best reagents in order to obtain the corresponding halohydrins **2** ( $\text{X}=\text{Cl}, \text{Br}, \text{I}$ ) under very mild reaction conditions (THF, room temperature; Table 1, entries 15, 16, and 18-20) acting both as Lewis acids and as a source of halogenide anions. Alternatively,  $\text{Ti}(\text{OPr})_4$  can be used when ethylmagnesium bromide or iodine<sup>6</sup> were used for the preparation of bromo or iodohydrins **2ah** or **2ai**.



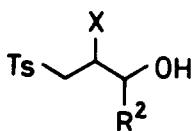
$\text{X} = \text{OMe}, \text{OPr}^t, \text{NEt}_2, \text{N}_3, \text{SCN}, \text{SPh}, \text{Hal}, \text{PPh}_3, \text{CN}.$

Scheme 1

In the case of soft nucleophiles such as sodium azide, triphenylphosphine,<sup>8</sup> and sodium benzenethiolate, the  $\gamma$ -ring-opening of epoxy sulfones **1** took place in the absence of  $\text{Ti}(\text{OPr})_4$  (Table 1, entries 6 and 7, 22, and 12-14, respectively). The corresponding phosphonium salt was isolated as its bromide **2aj** by treatment of the reaction mixture with concentrated hydrobromic acid.

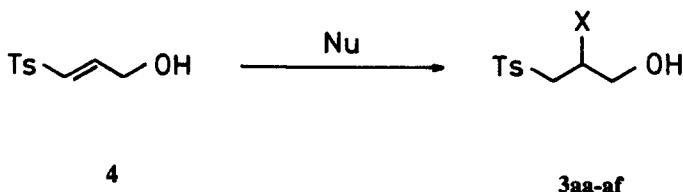


The regiochemistry of the process was in all cases the corresponding to the attack of the heteronucleophile at the  $\gamma$ -position of the  $\beta,\gamma$ -epoxy sulfone **1**, except in the reaction of sodium azide with compound **1a**, which afforded a 30% of compound **3ad**. The stereochemistry was studied in the reaction of *trans*-1-methyl-2-(tosylmethyl)oxirane (**1c**)<sup>9</sup> with different nucleophiles in the presence or not of  $\text{Ti}(\text{OPr})_4$  yielding compounds **2** as the corresponding *threo* stereoisomers (Table 1, entries 8, 14, and 19).



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When more basic reagents were used as nucleophiles the regioisomeric  $\beta$ -functionalized  $\gamma$ -hydroxy sulfones 3 were exclusively obtained. Sodium methoxide gave compound 3aa and diethylamine compounds 3ac and 3cc (Table 1, entries 1 and 3-5) with or without  $Ti(OPr^t)_4$ . Products 3 should arise from the Michael addition of the nucleophile to the  $\gamma$ -tosylated allylic alcohol formed under the more basic reaction conditions by  $\alpha$ -proton abstraction and  $\beta$ -elimination of the epoxy sulfone 1. This fact was confirmed in the reaction of 3-tosyl-2-propen-1-ol 4 with different nucleophiles such as sodium methoxide, diethylamine, or sodium azide to afford products 3aa, 3ac, or 3ad. Compounds 3ae and 3af were also prepared by reaction of allylic alcohol 4 with butylamine and sodium benzenethiolate, respectively (Scheme 2 and Table 2).



X = OMe, NEt<sub>2</sub>, N<sub>3</sub>, NHBu<sup>a</sup>, SPh.

Scheme 2

$\beta$ -Functionalized  $\gamma$ -hydroxy sulfones 3 are adequate precursors for  $\alpha$ -functionalized  $\alpha,\beta$ -unsaturated carbonyl compounds by oxidation of the alcohol function followed by *p*-toluenesulfinate elimination.<sup>2,10</sup> On the other hand, compound 3ad could be a good precursor of 2-boc-amino-3-phenylsulfonyl-1-(2-tetrahydropyranyloxy)propane, used as building block for  $\beta,\gamma$ -unsaturated  $\alpha$ -amino acids.<sup>11</sup>

From the results described in this paper we conclude that  $\beta,\gamma$ -epoxy sulfones are appropriate starting materials for the synthesis of  $\beta$ - and  $\gamma$ -functionalized  $\gamma$ - and  $\beta$ -hydroxy sulfones, respectively, by reaction with heteronucleophiles.

## EXPERIMENTAL

**General.** For general information see ref. 2. All NMR spectra were recorded on a Bruker AC-300 spectrometer in  $CDCl_3$  as solvent.  $^{13}C$  NMR assignments were done on the basis of DEPT experiments.

**Table 1.** Nucleophilic Opening of  $\beta,\gamma$ -Epoxy Sulfones.

Entry	Epoxide	Nu (equiv.)	Ti(O <i>Pr</i> <sup>1</sup> ) <sub>4</sub> (equiv.)	Reaction conditions				Product		
				Solvent	T <sup>a</sup> (°C)	Time	no.	X	Yield, <sup>a</sup> %	
1	1a	NaOMe (1.5)	0	MeOH	rt	5h	3aa	O <i>Me</i>	82	
2	1a	Pr <sup>1</sup> OH (exc.)	1.5	Pr <sup>1</sup> OH	reflux	3d	2ab	O <i>Pr</i> <sup>1</sup>	52	
3	1a	Et <sub>2</sub> NH (exc.)	1.5	Et <sub>2</sub> NH	rt	1d	3ac	NEt <sub>2</sub>	60	
4	1a	Et <sub>2</sub> NH (exc.)	0	Et <sub>2</sub> NH	reflux	1d	3ac	NEt <sub>2</sub>	65	
5	1c	Et <sub>2</sub> NH (exc.)	0	Et <sub>2</sub> NH	reflux	1d	3cc <sup>b</sup>	NEt <sub>2</sub>	56	
6	1a	NaN <sub>3</sub> (5) <sup>c</sup>	0	MeOH <sup>d</sup>	reflux	7h	2ad <sup>e</sup>	N <sub>3</sub>	80	
7	1b	NaN <sub>3</sub> (5) <sup>c</sup>	0	MeOH <sup>d</sup>	reflux	6h	2bd	N <sub>3</sub>	56	
8	1a	NH <sub>4</sub> SCN (1.5)	1.5	THF	rt	15min	2ae	SCN	82	
9	1b	NH <sub>4</sub> SCN (1.5)	1.5	THF	rt	30min	2be	SCN	75	
10	1c	NH <sub>4</sub> SCN (1.5)	1.5	THF	rt	30min	2ce	SCN	80	
11	1a	PhSH (1.5)	1.5	THF	reflux	1d	2af	SPh	86	
12	1a	NaSPh (1.2)	0	THF	rt	10min	2af	SPh	89	
13	1b	NaSPh (1.2)	0	THF	rt	15min	2bf	SPh	82	
14	1c	NaSPh (1.2)	0	THF	rt	30min	2cf	SPh	91	
15	1a	MgCl <sub>2</sub> (1.2)	0	THF	rt	20min	2ag	C1	89	
16	1a	MgBr <sub>2</sub> (1.2)	0	THF	rt	15min	2ah	Br	86	

Table 1. cont.

17	<b>1a</b>	EtMgBr (1.2)	1.5	THF	rt	20min	<b>2ab</b>	Br	87
18	<b>1b</b>	MgBr <sub>2</sub> (1.2)	0	THF	rt	15min	<b>2bh</b>	Br	83
19	<b>1c</b>	MgBr <sub>2</sub> (1.2)	0	THF	rt	15min	<b>2ch</b>	Br	75
20	<b>1a</b>	MgI <sub>2</sub> (1.2)	0	THF	rt	20min	<b>2ai</b>	I	92
21	<b>1a</b>	I <sub>2</sub> (1.1)	1.1	CH <sub>2</sub> Cl <sub>2</sub>	rt	25min	<b>2ai</b>	I	93
22	<b>1a</b>	PPh <sub>3</sub> (1.0)	0	PhOH	60	4h	<b>2aj</b> <sup>f</sup>	PPh <sub>3</sub>	74
23	<b>1a</b>	Me <sub>3</sub> SiCN (3)	1.5	DME	reflux	1d	<b>2ak</b>	CN	60

<sup>a</sup> Isolated yield after column chromatography (silica gel), based on the starting epoxide 1.<sup>b</sup> Erythro/threo mixture 1/2. (<sup>1</sup>H-NMR).<sup>c</sup> Two equivalents of ammonium chloride were added.<sup>d</sup> MeOH/H<sub>2</sub>O : 8/1 (<sup>1</sup>H-NMR).<sup>e</sup> Compound **3ad** was also obtained. Regioisomers ratio **2ad/3ad** : 7/3.<sup>f</sup> Isolated as the corresponding phosphonium bromide after treatment of the reaction mixture with conc. hydrobromic acid.

**Table 2.** Michael Addition of Heteronucleophiles to 3-Tosyl-2-propen-1-ol (**4**).

Entry	Nu (equiv.)	Reaction conditions			Time	no.	X	Yield, <sup>a</sup> %
		Solvent	T <sup>a</sup> (°C)					
1	NaOMe (1.5)	MeOH	rt		4h	<b>3aa</b>	OMe	96
2	Et <sub>2</sub> NH (exc.)	Et <sub>2</sub> NH	reflux		20h	<b>3ac</b>	NET <sub>2</sub>	84
3	NaN <sub>3</sub> (2.5)	MeOH	reflux		1d	<b>3ad</b>	N <sub>3</sub>	80
4	Bu <sup>n</sup> NH <sub>2</sub> (exc.)	Bu <sup>n</sup> NH <sub>2</sub>	reflux		6h	<b>3ae</b>	NHBu <sup>n</sup>	91
5	NaSPh (2.5)	THF	rt		3h	<b>3af</b>	SPh	70

<sup>a</sup> Isolated yield after column chromatography (silica gel), based on the starting compound **4**.

#### Nucleophilic Opening of $\beta,\gamma$ -Epoxy Sulfones. General Procedure.

**a) By means of Ti(OPr)<sub>4</sub>.** A solution of the epoxide **1** (0.5 mmol), nucleophile (see Table 1) and titanium(IV) isopropoxide (0.45 ml, 1.5 mmol) in the corresponding solvent (*ca.* 5ml) was stirred under argon at the temperature and for the time shown in Table 1. The resulting mixture was hydrolyzed with a 9% aqueous sulfuric acid solution and extracted with ether (2x10 ml). The organic layer was washed with water, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated in vacuo (15 torr). The resulting residue was purified by column chromatography on silica gel (hexane/ether) or/and by recrystallization to afford products **2**.

**b) Without Ti(OPr)<sub>4</sub>.** The solution of the epoxide **1** (0.5 mmol) and nucleophile<sup>12</sup> in the appropriate solvent (*ca.* 5 ml) was stirred under argon<sup>13</sup> in the conditions shown in Table 1. The reaction mixture was hydrolyzed with water and worked up as above<sup>14</sup> to yield products **2** or/and **3**.

**2-Methoxy-3-tosyl-1-propanol (3aa):** mp 240–241°C (dec., hexane/CH<sub>2</sub>Cl<sub>2</sub>);  $\nu_{\text{max}}$  (CDCl<sub>3</sub>) 3490 (OH), 1300, and 1140 cm<sup>-1</sup> (SO<sub>2</sub>);  $\delta_{\text{H}}$  2.43 (s, 3H, CH<sub>3</sub>Ar), 2.92 (br s, 1H, OH), 3.26 (s, 3H, CH<sub>3</sub>O), 3.37 (d, *J*=5.5Hz, 2H, CH<sub>2</sub>O), 3.56 (dd, *J*=11.6, 4.0Hz, 1H, 1xCH<sub>2</sub>S), 3.69–3.82 (m, 2H, CHO and 1xCH<sub>2</sub>S), 7.34, and 7.79 (2d, *J*=8.1Hz, 4H, ArH);  $\delta_{\text{C}}$  21.36 (CH<sub>3</sub>Ar), 57.06 (CH<sub>3</sub>O), 57.40 (CH<sub>2</sub>S), 62.28 (CH<sub>2</sub>O), 75.99 (CHO), 127.82, 129.59, 136.81, and 144.53 (ArC); *m/z* 213 (*M*<sup>+</sup>-31, 24%), 156 (14), 155 (95), 149 (19), 92 (20), 91 (100), 89 (12), 65 (21), and 59 (23). Anal. Calcd. for C<sub>11</sub>H<sub>16</sub>O<sub>4</sub>S:C, 54.08; H, 6.60. Found C, 53.75, H, 6.51.

**1-Isopropoxy-3-tosyl-2-propanol (2ab):** *R*<sub>f</sub> 0.6 (hexane/ether:1/10);  $\nu_{\text{max}}$  (film) 3450 (OH), 1250, and 1125 cm<sup>-1</sup> (SO<sub>2</sub>);  $\delta_{\text{H}}$  1.09, 1.12 (2d, *J*=5.6Hz, 6H, 2xCH<sub>3</sub>CH), 2.45 (s, 3H, CH<sub>3</sub>Ar), 3.15 (br s, 1H, OH), 3.30 (d, *J*=7.5Hz, 1H, 1xCH<sub>2</sub>S), 3.31 (d, *J*=4.0Hz, 1H, 1xCH<sub>2</sub>S), 3.45 (d, *J*=5.1Hz, 2H, CH<sub>2</sub>O), 3.56 (m, *J*=6.1Hz, CHCH<sub>3</sub>), 4.23 (m, 1H, CHCH<sub>2</sub>), 7.37, and 7.81 (2d, *J*=8.1Hz, 4H, ArH);  $\delta_{\text{C}}$  21.58, 21.87, 21.90 (3xCH<sub>3</sub>), 59.67 (CH<sub>2</sub>S), 65.65 (CH<sub>2</sub>O), 70.33, 72.34 (2xCHO), 128.00, 129.92, 136.50, and 144.92 (ArC); *m/z* 254 (*M*<sup>+</sup>-18, 1.6%), 199 (19), 157 (12), 156 (23), 155 (43), 92 (71), 91 (100), 89 (16), 65 (45), 63 (17), 49 (16), 43 (37), and 41 (13).

**2-Diethylamino-3-tosyl-1-propanol (3ac):**  $R_f$  0.4 (hexane/ether:1/10);  $\nu_{\max}$  (film) 3400 (OH), 1300, and 1140  $\text{cm}^{-1}$  ( $\text{SO}_2$ );  $\delta_{\text{H}}$  0.9 (t,  $J=7.1\text{Hz}$ , 6H, 2x $\text{CH}_3\text{CH}_2$ ), 2.13, 2.17, 2.44, 2.48 (4q,  $J=7.0\text{Hz}$ , 4H, 2x $\text{CH}_2\text{N}$ ), 2.38 (s, 3H,  $\text{CH}_3\text{Ar}$ ), 2.70-2.85 (m with dd at 2.81,  $J=14.0$ , 9.0Hz, 2H, CHN and 1x $\text{CH}_2\text{S}$ ), 3.16 (dd,  $J=14.0$ , 1.7Hz, 1H, 1x $\text{CH}_2\text{S}$ ), 3.27-3.57 (m, 2H, 1x $\text{CH}_2\text{O}$  and OH), 3.57 (d,  $J=5.0\text{Hz}$ , 1H, 1x $\text{CH}_2\text{O}$ ), 7.30, and 7.72 (2d,  $J=8.2\text{Hz}$ , 4H, ArH);  $\delta_{\text{C}}$  13.99 ( $\text{CH}_3\text{CH}_2$ ), 21.56 ( $\text{CH}_3\text{Ar}$ ), 43.22 ( $\text{CH}_2\text{N}$ ), 53.27 (CHN), 55.53 ( $\text{CH}_2\text{S}$ ), 61.14 ( $\text{CH}_2\text{O}$ ), 127.85, 130.02, 136.39, and 144.95 (ArC);  $m/z$  285 ( $M^+$ , 0.2%), 254 (56), 116 (18), 114 (100), 100 (21), 99 (100), 92 (12), 91 (78), 86 (18), 84 (66), 72 (28), 71 (93), 70 (56), 65 (55), 58 (31), 57 (20), 56 (93), 44 (24), 43 (14), and 42 (55).

(threo/erythro)-3-Diethylamino-4-tosyl-2-butanol (3cc):  $R_f$  0.5 (hexane/ether:1/10);  $\nu_{\max}$  (film) 3450 (OH), 1300, and 1140  $\text{cm}^{-1}$  ( $\text{SO}_2$ );  $\delta_{\text{H}}$  0.96, 1.07 (2t,  $J=7.1\text{Hz}$ , 6H, 2x $\text{CH}_3\text{CH}_2$ ), 1.14, 1.28 (2d,  $J=6.0\text{Hz}$ , 3H,  $\text{CH}_3\text{CH}$ ), 2.20-2.60 (m with s at 2.45, 8H, 2x $\text{CH}_2\text{CH}_3$ , OH, and  $\text{CH}_3\text{Ar}$ ), 2.84-3.75 (6m, 4H,  $\text{CH}_2\text{S}$ , CHN, and CHO), 7.36, and 7.80 (2d,  $J=8.1\text{Hz}$ , 4H, ArH);  $\delta_{\text{C}}$  14.40 ( $\text{CH}_3\text{CH}_2$ ), 18.76 ( $\text{CH}_3\text{CO}$ ), 21.53 ( $\text{CH}_3\text{Ar}$ ), 44.21 ( $\text{CH}_2\text{N}$ ), 53.52 (CHN), 60.39 ( $\text{CH}_2\text{S}$ ), 66.24 (CHO), 127.65, 129.84, 136.92, and 144.82 (ArC);  $m/z$  299 ( $M^+$ , 0.4%), 255 (15), 254 (100), 114 (34), 99 (21), 91 (14), and 84 (10).

3-Azido-1-tosyl-2-propanol (2ad) and 2-azido-3-tosyl-1-propanol (3ad):  $R_f$  0.5 (hexane/ether:1/10);  $\nu_{\max}$  (film) 3500 (OH), 2100 ( $\text{N}_3$ ), 1290, and 1140  $\text{cm}^{-1}$  ( $\text{SO}_2$ );  $\delta_{\text{H}}$  (for 2ad) 2.45 (s, 3H,  $\text{CH}_3$ ), 3.20-3.43 (m, 5H,  $\text{CH}_2\text{S}$ ,  $\text{CH}_2\text{N}$ , and OH), 4.32 (m, 1H, CHO), 7.37, and 7.80 (2d,  $J=8.1\text{Hz}$ , 4H, ArH);  $\delta_{\text{C}}$  (for 2ad) 21.53 ( $\text{CH}_3$ ), 55.23 ( $\text{CH}_2\text{N}$ ), 59.48 ( $\text{CH}_2\text{S}$ ), 65.58 (CHO), 127.82, 130.07, 135.92, and 145.35 (ArC);  $m/z$  199 ( $M^+$ -56, 19%), 155 (43), 92 (30), 91 (100), 89 (12), 65 (38), 63 (12), 44 (11), and 43 (18).

3-Azido-2-methyl-1-tosyl-2-propanol (2bd):  $R_f$  0.5 (hexane/ether:1/5);  $\nu_{\max}$  (film) 3500 (OH), 2110 ( $\text{N}_3$ ), 1300, and 1150  $\text{cm}^{-1}$  ( $\text{SO}_2$ );  $\delta_{\text{H}}$  1.45 (s, 3H,  $\text{CH}_3\text{CO}$ ), 2.45 (s, 3H,  $\text{CH}_3\text{Ar}$ ), 3.26, 3.40 (2d,  $J=14.2\text{Hz}$ , 2H,  $\text{CH}_2\text{S}$ ), 3.39, 3.48 (2d,  $J=12.4\text{Hz}$ , 2H,  $\text{CH}_2\text{N}$ ), 3.87 (br s, 1H, OH), 7.37, and 7.80 (2d,  $J=8.1\text{Hz}$ , 4H, ArH);  $\delta_{\text{C}}$  21.43 ( $\text{CH}_3\text{Ar}$ ), 25.31 ( $\text{CH}_3\text{CO}$ ), 59.57 ( $\text{CH}_2\text{S}$ ), 62.22 ( $\text{CH}_2\text{N}$ ), 72.18 (CO), 127.43, 129.90, 137.57, and 145.07 (ArC);  $m/z$  213 ( $M^+$ -56, 66%), 157 (10), 155 (60), 139 (14), 105 (12), 92 (15), 91 (100), 89 (14), 77 (12), 65 (39), and 43 (23).

(2-Hydroxy-3-tosyl)propyl thiocyanate (2ae)<sup>15</sup>:  $R_f$  0.45 (hexane/ether:1/10);  $\nu_{\max}$  (film) 3440 (OH), 2140 (SCN), 1290, and 1135  $\text{cm}^{-1}$  ( $\text{SO}_2$ );  $\delta_{\text{H}}$  2.44 (s, 3H,  $\text{CH}_3$ ), 3.02 (dd,  $J=13.5$ , 7.0Hz, 1H, 1x $\text{CH}_2\text{SCN}$ ), 3.18 (dd,  $J=13.5$ , 4.0Hz, 1H, 1x $\text{CH}_2\text{S}$ ), 3.38-3.43 (m, 2H,  $\text{CH}_2\text{SO}_2$ ), 4.10 (br s, 1H, OH), 4.48 (m, 1H, CHO), 7.37, and 7.80 (2d,  $J=8.0\text{Hz}$ , 4H, ArH);  $\delta_{\text{C}}$  21.48 ( $\text{CH}_3$ ), 39.47 ( $\text{CH}_2\text{SCN}$ ), 60.09 ( $\text{CH}_2\text{SO}_2$ ), 65.12 (CHO), 112.24 (SCN), 127.78, 130.06, 135.72, and 145.42 (ArC);  $m/z$  271 ( $M^+$ , 0.5%), 199 (14), 155 (47), 139 (17), 92 (21), 91 (100), 89 (18), 77 (10), 73 (23), 72 (16), 65 (41), 63 (17), 46 (15), and 45 (28).

(2-Hydroxy-2-methyl-3-tosyl)propyl thiocyanate (2be):<sup>15</sup>  $R_f$  0.4 (hexane/ether:1/10);  $\nu_{\max}$  (film) 3480 (OH), 2140 (SCN), 1300, and 1150  $\text{cm}^{-1}$  ( $\text{SO}_2$ );  $\delta_{\text{H}}$  1.52 (s, 3H,  $\text{CH}_3\text{CO}$ ), 2.46 (s, 3H,  $\text{CH}_3\text{Ar}$ ), 2.60 (s, 2H,  $\text{CH}_2\text{SCN}$ ), 3.04, 3.51 (2d,  $J=14.1\text{Hz}$ , 2H,  $\text{CH}_2\text{SO}_2$ ), 3.40 (br s, 1H, OH), 7.37, and 7.79 (2d,  $J=8.0\text{Hz}$ , 4H, ArH);  $\delta_{\text{C}}$  21.33, 21.57 (2x $\text{CH}_3$ ), 51.97, 52.88 (2x $\text{CH}_2$ ), 63.68 (CO), 112.75 (SCN), 127.96, 129.93, 136.86, and 145.05 (ArC);  $m/z$  270 ( $M^+$ -15, 0.1%), 213 (46), 155 (51), 139 (16), 105 (14), 92 (19), 91 (100), 89 (20), 87 (38), 77 (12), 65 (53), 63 (16), 59 (12), 53 (11), 46 (13), 45 (24), and 43 (21).

(threo)-(2-Hydroxy-1-methyl-3-tosyl)propyl thiocyanate (2ce):<sup>15</sup>  $R_f$  0.3 (hexane/ether:1/10);  $\nu_{\max}$  (film) 3520 (OH), 2130 (SCN), 1290, and 1130  $\text{cm}^{-1}$  ( $\text{SO}_2$ );  $\delta_{\text{H}}$  1.49 (d,  $J=6.9\text{Hz}$ , 3H,  $\text{CH}_3\text{CH}$ ), 2.47 (s, 3H,  $\text{CH}_3\text{Ar}$ ), 3.34 (d,  $J=5.6\text{Hz}$ , 2H,  $\text{CH}_2\text{S}$ ), 3.45 (m, 1H, CHSCN), 3.58 (br s, 1H, OH), 4.37 (m, 1H, CHO), 7.41, and 7.82 (2d,  $J=8.0\text{Hz}$ , 4H, ArH);  $\delta_{\text{C}}$  16.03 ( $\text{CH}_3\text{CH}$ ), 21.84 ( $\text{CH}_3\text{Ar}$ ), 48.52 (CHSCN), 53.39 ( $\text{CH}_2\text{SO}_2$ ), 68.67 (CHO), 110.74 (SCN), 127.78, 130.07, 135.65, and 145.42 (ArC);  $m/z$  286 ( $M^+$ +1, 1%), 285 ( $M^+$ , 4), 199 (81), 157 (11), 156 (11), 155 (97), 139 (15), 112 (10), 92 (23), 91 (100), 89 (15), 87 (19), 65 (40), and 59 (12).

1-Phenylthio-3-tosyl-2-propanol (2af): mp 111-112 °C (hexane/ $\text{CH}_2\text{Cl}_2$ );  $\nu_{\max}$  ( $\text{CHCl}_3$ ) 3500(OH), 1300, and 1150  $\text{cm}^{-1}$  ( $\text{SO}_2$ );  $\delta_{\text{H}}$  2.33 (s, 3H,  $\text{CH}_3$ ), 2.88 (dd,  $J=14.0$ , 6.6Hz, 1H, 1x $\text{CH}_2\text{SPh}$ ), 2.99 (dd,  $J=14.0$ ,

6.0Hz, 1H, 1xCH<sub>2</sub>SPh), 3.19 (dd, *J*=14.4, 8.6Hz, 1H, 1xCH<sub>2</sub>SO<sub>2</sub>), 3.31 (br s, 1H, OH), 3.41 (dd, *J*=14.4, 2.6Hz, 1H, 1xCH<sub>2</sub>SO<sub>2</sub>), 4.04 (m, 1H, CHO), 7.10-7.25 (m with d at 7.21, *J*=8.2Hz, 7H, Ph and 2xp-TolH), and 7.64 (d, *J*=8.2Hz, 2H, *p*-TolH); δ<sub>C</sub> 21.47 (CH<sub>3</sub>), 39.91 (CH<sub>2</sub>SPh), 60.56 (CH<sub>2</sub>SO<sub>2</sub>), 65.05 (CHO), 126.73, 127.80, 128.90, 128.92, 130.19, 134.17, 135.87, and 144.89 (ArC); *m/z* 322 (*M*<sup>+</sup>, 9%), 155 (21), 149 (90), 124 (16), 123 (45), 116 (11), 109 (15), 92 (10), 91 (100), 89 (13), 79 (13), 77 (31), and 65 (57). Anal. Calcd. for C<sub>16</sub>H<sub>18</sub>O<sub>3</sub>S<sub>2</sub>: C, 59.60; H, 5.63. Found: C, 60.04; H, 5.67.

**1-Phenylthio-2-methyl-3-tosyl-2-propanol (2bf):** *R*, 0.5 (hexane/ether:1/10); ν<sub>max</sub> (film) 3480 (OH), 1300, and 1140 cm<sup>-1</sup> (SO<sub>2</sub>); δ<sub>H</sub> 1.48 (s, 3H, CH<sub>3</sub>CO), 2.39 (s, 3H, CH<sub>3</sub>Ar), 3.29 (d, *J*=3.8Hz, 2H, CH<sub>2</sub>SPh), 3.37, 3.50 (2d, *J*=14.3Hz, 2H, CH<sub>2</sub>SO<sub>2</sub>), 3.86 (s, 1H, OH), 7.15-7.36 (m with d at 7.28, *J*=8.2Hz, Ph and 2xp-TolH), and 7.73 (d, *J*=8.2Hz, 2H, 2xp-TolH); δ<sub>C</sub> 21.35 (CH<sub>3</sub>Ar), 26.82 (CH<sub>3</sub>CO), 45.70 (CH<sub>2</sub>SPh), 63.29 (CH<sub>2</sub>SO<sub>2</sub>), 72.38 (CO), 126.18, 127.34, 128.76, 129.43, 129.69, 135.99, 137.59, and 144.68 (ArC); *m/z* 338 (*M*<sup>+</sup>+2, 1%), 337 (*M*<sup>+</sup>+1, 1), 336 (*M*<sup>+</sup>, 7), 213 (84), 163 (17), 155 (77), 139 (10), 125 (10), 124 (100), 123 (36), 91 (84), 77 (19), and 65 (26).

(*threo*)-3-*Phenylthio-1-tosyl-2-butanol (2cf):* mp 86-87°C(hexane/ether); ν<sub>max</sub> (CDCl<sub>3</sub>) 3500 (OH), 1290, and 1130 cm<sup>-1</sup> (SO<sub>2</sub>); δ<sub>H</sub> 1.17 (d, *J*=7.0Hz, 3H, CH<sub>3</sub>CH), 2.34 (s, 3H, CH<sub>3</sub>Ar), 3.22 (m, 1H, CH<sub>2</sub>SPh), 3.30 (dd, *J*=14.4, 8.9Hz, 1H, 1xCH<sub>2</sub>S), 3.40 (d, *J*=2.6Hz, 1H, OH), 3.52 (dd, *J*=14.4, 2.3Hz, 1H, 1xCH<sub>2</sub>S), 4.06 (m, 1H, CHO), 7.14-7.40 (m with d at 7.22, *J*=8.1Hz, 7H, Ph and 2xp-TolH), and 7.65 (d, *J*=8.1Hz, 2H, *p*-TolH); δ<sub>C</sub> 15.40 (CH<sub>3</sub>CS), 21.49 (CH<sub>3</sub>Ar), 48.17 (CHS), 59.78 (CH<sub>2</sub>S), 67.93 (CHO), 127.53, 127.86, 128.88, 129.86, 132.71, 135.93, and 144.88 (ArC); *m/z* 338 (*M*<sup>+</sup>+2, 1%), 337 (*M*<sup>+</sup>+1, 1), 336 (*M*<sup>+</sup>, 6), 163 (31), 155 (13), 138 (15), 137 (100), 135 (17), 109 (23), 91 (39), and 65 (18). Anal. Calcd. for C<sub>17</sub>H<sub>20</sub>O<sub>3</sub>S<sub>2</sub>: C, 60.68; H, 5.99. Found: C, 60.31; H, 5.87.

**3-Chloro-1-tosyl-2-propanol (2ag):** mp 69-70°C (hexane/ether); ν<sub>max</sub> (CDCl<sub>3</sub>) 3480 (OH), 1290, and 1135 cm<sup>-1</sup> (SO<sub>2</sub>); δ<sub>H</sub> 2.37 (s, 3H, CH<sub>3</sub>), 3.20-3.31 (m, 2H, CH<sub>2</sub>S), 3.53 (d, *J*=4.9Hz, 2H, CH<sub>2</sub>Cl), 3.80 (br s, 1H, OH), 4.30 (m, 1H, CHO), 7.29, and 7.73 (2d, *J*=7.8Hz, 4H, ArH); δ<sub>C</sub> 21.51 (CH<sub>3</sub>), 47.56 (CH<sub>2</sub>Cl), 59.51 (CH<sub>2</sub>S), 66.10 (CHO), 127.85, 130.00, 135.99, and 145.25 (ArC); *m/z* 250 (*M*<sup>+</sup>+2, 1%), 248 (*M*<sup>+</sup>, 2), 199 (42), 157 (10), 155 (81), 139 (11), 92 (71), 91 (100), 89 (10), and 65 (24). Anal. Calcd. for C<sub>10</sub>H<sub>13</sub>ClO<sub>3</sub>S: C, 48.30; H, 5.27. Found: C, 48.56; H, 5.48.

**3-Bromo-1-tosyl-2-propanol (2ah):** mp 67-68°C (hexane/ether); ν<sub>max</sub> (CDCl<sub>3</sub>) 3450 (OH), 1300, and 1140 cm<sup>-1</sup> (SO<sub>2</sub>); δ<sub>H</sub> 2.44 (s, 3H, CH<sub>3</sub>), 3.30-3.55 (m, 5H, CH<sub>2</sub>S, CH<sub>2</sub>Br, and OH), 4.35 (m, 1H, CHO), 7.37, 7.81 (2d, *J*=8.0Hz, 4H, ArH); δ<sub>C</sub> 21.49 (CH<sub>3</sub>), 36.6 (CH<sub>2</sub>Br), 60.29 (CH<sub>2</sub>S), 65.55 (CHO), 127.81, 129.95, 135.95, and 145.18 (ArC); *m/z* 294 (*M*<sup>+</sup>+2, 1%), 292 (*M*<sup>+</sup>, 1), 199 (31), 155 (39), 139 (21), 105 (13), 92 (73), 91 (100), 89 (18), 77 (12), 65 (53), 57 (10), and 43 (17). Anal. Calcd. for C<sub>10</sub>H<sub>13</sub>BrO<sub>3</sub>S: C, 40.96; H, 4.44. Found: C, 41.36; H, 4.57.

**3-Bromo-2-methyl-1-tosyl-2-propanol (2bh):** *R*, 0.45 (hexane/ether:1/5); ν<sub>max</sub> (film) 3460 (OH), 1300, and 1140 cm<sup>-1</sup> (SO<sub>2</sub>); δ<sub>H</sub> 1.48 (s, 3H, CH<sub>3</sub>CO), 2.37 (s, 3H, CH<sub>3</sub>Ar), 3.22, 3.45 (2d, *J*=14.3Hz, 2H, CH<sub>2</sub>S), 3.49, 3.55 (2d, *J*=10.0Hz, 2H, CH<sub>2</sub>Br), 3.77 (br s, 1H, OH), 7.29, and 7.73 (2d, *J*=8.1Hz, 4H, ArH); δ<sub>C</sub> 21.51 (CH<sub>3</sub>Ar), 26.09 (CH<sub>3</sub>CO), 41.43 (CH<sub>2</sub>Br), 62.43 (CH<sub>2</sub>S), 71.27 (CO), 127.55, 129.94, 137.56, and 145.08 (ArC); *m/z* 308 (*M*<sup>+</sup>+2, 0.7%), 306 (*M*<sup>+</sup>, 0.8), 214 (11), 213 (100), 155 (46), 139 (10), 105 (10), 92 (11), 91 (42), and 65 (17).

(*threo*)-3-Bromo-1-tosyl-2-butanol (2ch): mp 75-76°C (hexane/ether); ν<sub>max</sub> (CDCl<sub>3</sub>) 3460 (OH), 1300, and 1140 cm<sup>-1</sup> (SO<sub>2</sub>); δ<sub>H</sub> 1.65 (d, *J*=7.0Hz, 3H, CH<sub>3</sub>CH), 2.45 (s, 3H, CH<sub>3</sub>Ar), 3.33 (dd, *J*=14.4, 8.8Hz, 1H, 1xCH<sub>2</sub>S), 3.50 (dd, *J*=14.4, 1.7Hz, 1H, 1xCH<sub>2</sub>S), 3.64 (d, *J*=3.8Hz, 1H, OH), 4.16 (m, 2H, CHO and CHBr), 7.38, and 7.82 (2d, *J*=8.1Hz, 4H, ArH); δ<sub>C</sub> 21.00 (CH<sub>3</sub>CH), 21.49 (CH<sub>3</sub>Ar), 52.15 (CHBr), 59.62 (CH<sub>2</sub>), 70.16 (CHO), 127.83, 129.93, 135.98, and 145.12 (ArC); *m/z* 308 (*M*<sup>+</sup>+2, 0.3%), 306 (*M*<sup>+</sup>, 0.4), 200 (10), 199 (97), 157 (12), 156 (10), 155 (99), 139 (16), 107 (13), 105 (11), 92 (41), 91 (100), 89 (15), 65 (45), 63 (14), and 43 (15). Anal. Calcd. for C<sub>11</sub>H<sub>15</sub>BrO<sub>3</sub>S: C, 43.01; H, 4.92. Found: C, 43.00; H, 5.10.

**3-Iodo-1-tosyl-2-propanol (2ai):** mp 115–116°C (hexane/CH<sub>2</sub>Cl<sub>2</sub>);  $\nu_{\text{max}}$  (CDCl<sub>3</sub>) 3480 (OH), 1300, and 1140 cm<sup>-1</sup> (SO<sub>2</sub>);  $\delta_{\text{H}}$  2.45 (s, 3H, CH<sub>3</sub>), 3.28–3.42 (m, 5H, CH<sub>2</sub>I, CH<sub>2</sub>S, and OH), 4.09 (m, 1H, CHO), 7.38, and 7.81 (2d,  $J=8.0$ Hz, 4H, ArH);  $\delta_{\text{C}}$  11.31 (CH<sub>2</sub>I), 21.53 (CH<sub>3</sub>), 61.75 (CH<sub>2</sub>S), 65.34 (CHO), 127.82, 129.96, 135.92, and 145.17 (ArC);  $m/z$  340( $M^+$ , 0.2%), 213 (49), 195 (26), 167 (16), 155 (53), 141 (11), 139 (86), 92 (19), 91 (100), 65 (32), and 63 (11). Anal. Calcd. for C<sub>11</sub>H<sub>15</sub>IO<sub>3</sub>S: C, 35.31; H, 3.85. Found: C, 36.00; H, 3.80.

**(2-Hydroxy-3-tosyl)propyltriphenylphosphonium bromide (2aj):** mp 127–128°C (hexane/CH<sub>2</sub>Cl<sub>2</sub>);  $\nu_{\text{max}}$  (Nujol) 3450 (OH), 1275, and 1130 cm<sup>-1</sup> (SO<sub>2</sub>);  $\delta_{\text{H}}$  2.39 (s, 3H, CH<sub>3</sub>), 3.76 (dt,  $J_{\text{gen}}=14.1$ Hz,  $J_{\text{P},\text{H}}=4.0$ Hz, 1H, 1xCH<sub>2</sub>P), 3.92 (dd,  $J=14.1$ , 13.4Hz, 1H, CH<sub>2</sub>S), 4.18 (dd,  $J=14.1$ , 8.2Hz, 1H, CH<sub>2</sub>S), 4.50–4.70 (m, 2H, CHO and 1xCH<sub>2</sub>P), 7.29 (d,  $J=8.0$ Hz, 2H, 2xp-TolH), 7.60–7.85 (m, 18H, 3xPh, 2xp-TolH, and OH);  $\delta_{\text{P}}$  58.99;  $\delta_{\text{C}}$  21.55 (CH<sub>3</sub>), 30.60 (d,  $J=53.0$ Hz, CH<sub>2</sub>P), 61.52 (d,  $J=14.9$ Hz, CH<sub>2</sub>S), 61.83 (d,  $J=4.7$ Hz, CHO), 118.32 (d,  $J=86.3$ Hz, C<sub>p</sub>, Ph), 127.73, 129.92, 136.09, 144.98 (p-TolC), 130.19 (d,  $J=12.7$ Hz, C<sub>m</sub>, Ph), 133.96 (d,  $J=10.3$ Hz, C<sub>c</sub>, Ph), and 134.90 (d,  $J=3.0$ Hz, C<sub>p</sub>, Ph);  $m/z$  386 ( $M^+$ –169, 5%), 278 (36), 277 (94), 201 (15), 199 (18), 183 (19), 155 (45), 152 (12), 139 (11), 132 (10), 92 (15), 91 (100), 89 (10), 77 (27), 65 (27), and 51 (18). Anal. Calcd. for C<sub>28</sub>H<sub>28</sub>BrO<sub>3</sub>PS: C, 60.54; H, 5.08. Found: C, 59.98; H, 5.08.

**2-Hydroxy-3-tosylbutanenitrile (2ak):** mp 100–101°C (hexane/ether);  $\nu_{\text{max}}$  (Nujol) 3400 (OH), 2240 (CN), 1275, and 1135 cm<sup>-1</sup> (SO<sub>2</sub>);  $\delta_{\text{H}}$  2.47 (s, 3H, CH<sub>3</sub>), 2.64 (dd,  $J=16.9$ , 5.9Hz, 1H, 1xCH<sub>2</sub>CN), 2.72 (dd,  $J=16.9$ , 5.6Hz, 1H, 1xCH<sub>2</sub>CN), 3.31 (dd,  $J=14.1$ , 3.4Hz, 1H, 1xCH<sub>2</sub>S), 3.39 (dd,  $J=14.1$ , 8.3Hz, 1H, 1xCH<sub>2</sub>S), 3.88 (br s, 1H, OH), 4.48 (m, 1H, CHO), 7.40, and 7.81 (2d,  $J=8.2$ Hz, 4H, ArH);  $\delta_{\text{C}}$  21.63 (CH<sub>3</sub>), 25.36 (CH<sub>2</sub>CN), 60.68 (CH<sub>2</sub>S), 62.50 (CHO), 116.18 (CN), 127.94, 130.27, 135.69, and 145.75 (ArC);  $m/z$  239 ( $M^+$ , 3%), 172 (10), 155 (46), 92 (42), 91 (100), 86 (11), 84 (21), and 65 (23). Anal. Calcd. for C<sub>11</sub>H<sub>13</sub>NO<sub>3</sub>S: C, 55.21; H, 5.48; N, 5.85. Found: C, 54.91; H, 5.40; N, 5.81.

**Michael addition of nucleophiles to 3-tosyl-2-propen-1-ol (4).** General procedure. A solution of compound 4 (106 mg, 0.5 mmol) and the nucleophile (see Table 2) in the corresponding solvent (5 ml) was stirred at the temperature indicated in Table 2. The reaction mixture was treated with water and extracted with ether (2x10 ml). The organic layer was dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated in vacuo (15 and 0.1 torr) to give crude products 3, which were purified by column chromatography (silica gel, hexane/ether) or/and by recrystallization.

**2-Azido-3-tosyl-1-propanol (3ad):**  $R_f$  0.5 (hexane/ether: 1/10);  $\nu_{\text{max}}$  (film) 3500 (OH), 2140 (N<sub>3</sub>), 1290, and 1130 cm<sup>-1</sup> (SO<sub>2</sub>);  $\delta_{\text{H}}$  2.46 (s, 3, CH<sub>3</sub>), 3.02 (s, 1H, OH), 3.25 (dd,  $J=14.6$ , 7.8Hz, 1H, 1xCH<sub>2</sub>S), 3.37 (dd,  $J=14.6$ , 4.4, 1H, 1xCH<sub>2</sub>S), 3.73, 3.74 (2d,  $J=5$ Hz, 2H, CH<sub>2</sub>O), 3.96 (m, 1H, CHN), 7.37, and 7.80 (2d,  $J=8.1$ Hz, 4H, ArH);  $\delta_{\text{C}}$  21.61 (CH<sub>3</sub>), 56.62 (CH<sub>2</sub>S), 57.94 (CHN), 64.10 (CH<sub>2</sub>O), 127.94, 130.04, 136.24, and 145.16 (ArC);  $m/z$  227 ( $M^+$ –28, 0.3%), 212 ( $M^+$ –43, 2), 156 (10), 155 (25), 139 (36), 133 (40), 132 (16), 105 (24), 92 (28), 91 (100), 89 (10), 77 (10), and 65 (30).

**2-Butylamino-3-tosyl-1-propanol (3ae):** mp 73–74°C (hexane/ether);  $\nu_{\text{max}}$  (CDCl<sub>3</sub>) 3500 (OH), 1300, and 1145 cm<sup>-1</sup> (SO<sub>2</sub>);  $\delta_{\text{H}}$  0.81 (t,  $J=7.0$ Hz, 3H, CH<sub>3</sub>CH<sub>2</sub>), 1.27 (m, 4H, CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 2.36 (s, 3H, CH<sub>3</sub>Ar), 2.44 (t,  $J=6.6$ Hz, 2H, CH<sub>2</sub>N), 2.59 (br s, 2H, NH and OH), 3.08 (m, 1H, CHN), 3.14 (m, 2H, CH<sub>2</sub>S), 3.37, 3.60 (2dd,  $J=11.1$ , 4.1Hz, 2H, CH<sub>2</sub>O), 7.28, and 7.71 (2d,  $J=8.0$ Hz, 4H, ArH);  $\delta_{\text{C}}$  13.73 (CH<sub>2</sub>CH<sub>2</sub>), 20.13 (CH<sub>2</sub>CH<sub>3</sub>), 21.42 (CH<sub>3</sub>Ar), 31.94 (CH<sub>2</sub>CH<sub>2</sub>N), 46.27 (CH<sub>2</sub>N), 54.26 (CHN), 57.41 (CH<sub>2</sub>S), 63.32 (CH<sub>2</sub>O), 127.65, 129.84, 136.4, and 144.78 (ArC);  $m/z$  255 ( $M^+$ –30, 15%), 254 (100), 155 (15), 139 (25), 114 (18), 91 (93), 89 (13), 84 (18), 65 (36), 57 (19), and 56 (48). Anal. Calcd. for C<sub>14</sub>H<sub>23</sub>NO<sub>3</sub>S: C, 58.92; H, 8.12; N, 4.91. Found: C, 59.29; H, 8.02; N, 4.85.

**2-Phenylthio-3-tosyl-1-propanol (3af):**  $R_f$  0.45 (hexane/ether: 1/10);  $\nu_{\text{max}}$  (film) 3500 (OH), 1300, and 1140 cm<sup>-1</sup> (SO<sub>2</sub>);  $\delta_{\text{H}}$  2.42 (s, 4H, CH<sub>3</sub> and OH), 3.30 (dd,  $J=14.1$ , 3.5Hz, 1H, 1xCH<sub>2</sub>SO<sub>2</sub>), 3.54 (dd,  $J=14.1$ , 8.5Hz, 1H, 1xCH<sub>2</sub>SO<sub>2</sub>), 3.63 (m, 1H, CHS), 3.84 (m, 2H, CH<sub>2</sub>O), 7.20–7.32 (m, 7H, Ph and 2xp-TolH), and 7.72 (d,  $J=8.2$ Hz, 2H, 2xp-TolH);  $\delta_{\text{C}}$  21.55 (CH<sub>3</sub>), 45.13 (CHS), 57.16 (CH<sub>2</sub>S), 62.82 (CH<sub>2</sub>O), 127.89, 128.02, 129.20, 129.94, 131.87, 132.66, 135.97, and 144.99 (ArC);  $m/z$  324 ( $M^+$  + 2, 1%), 323

( $M^+ + 1$ , 2), 322 ( $M^+$ , 12), 167 (24), 166 (35), 155 (18), 149 (22), 137 (11), 136 (62), 135 (52), 122 (100), 110 (14), 109 (27), 91 (64), 77 (13), and 65 (33).

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#### REFERENCES AND NOTES

1. Conrad, P. C.; Fuchs, P. L. *J. Am. Chem. Soc.* **1978**, *100*, 346.
2. Nájera, C.; Sansano, J. M. *Tetrahedron* **1990**, *46*, 3993.
3. (a) Kocienski, P. J. *Tetrahedron Lett.* **1979**, 441. (b) Eisch, J. J.; Galle, J. E. *J. Org. Chem.* **1979**, *44*, 3277. (c) Donaldson, R. E.; Fuchs, P. L. *J. Am. Chem. Soc.* **1981**, *103*, 2108. (d) Saddler, J. C.; Donaldson, R. E.; Fuchs, P. L. *J. Am. Chem. Soc.* **1981**, *103*, 2110. (e) Bäckvall, J.-E.; Juntunen, S. *K. J. Org. Chem.* **1988**, *53*, 2398. (f) Nájera, C.; Yus, M. *J. Org. Chem.* **1989**, *54*, 1491.
4. (a) Tanikaga, R.; Hosoya, K.; Kaji, A. *J. Chem. Soc., Chem. Commun.* **1986**, 836. (b) Tanikaga, R.; Hosoya, K.; Kaji, A. *J. Chem. Soc., Perkin Trans. 1* **1987**, 1799.
5. Caron, M.; Sharpless, K. B. *J. Org. Chem.* **1985**, *50*, 1557.
6. Alvarez, E.; Nuñez, M. T.; Martin, V. S. *J. Org. Chem.* **1990**, *55*, 3429.
7. Otsubo, K.; Inanaga, J.; Yamaguchi, M. *Tetrahedron Lett.* **1987**, *28*, 4435.
8. Christol, H.; Grelet, D.; Darvich, M. R.; Fallouh, F.; Plenat, F.; Cristau, H.-J. *Bull. Soc. Chim. Fr.* **1989**, 477.
9. Kocienski, P. J. *J. Chem. Soc., Perkin Trans. 1* **1983**, 945.
10. Nájera, C.; Mancheño, B.; Yus, M. *Tetrahedron Lett.* **1989**, *30*, 6085.
11. Sasaki, N. A.; Hashimoto, C.; Pauly, R. *Tetrahedron Lett.* **1989**, *30*, 1943.
12. Sodium thiophenolate was prepared from thiophenol and sodium methoxide, and magnesium iodide from magnesium and iodine in THF.
13. Except when sodium azide, diethylamine, and triphenylphosphine were used as nucleophiles.
14. In the case of the reaction with triphenylphosphine (Table 1, entry 22) 48% hydrobromic acid (1 mmol) was added and after 1 h stirring the reaction mixture was diluted with water.
15. Decomposed on standing.